

Ph.D. Physics

SYLLABUS

(Effective from the Academic year 2024 – 2025 and onwards)



DEPARTMENT OF PHYSICS

KARPAGAM ACADEMY OF HIGHER EDUCATION

(Deemed to be University Established Under Section 3 of UGC Act, 1956)

Eachanari Post, Coimbatore – 641 021, INDIA.

Phone: 0422-6453777, 6471113-5, 2980011-2980018;

Fax No: 0422 – 2980022, 2980023

Email: info@karpagam.com

Web: www.kahedu.edu.in

DEPARTMENT OF PHYSICS
FACULTY OF ARTS, SCIENCE, CCOMMERCE & MANAGEMENT
RESEARCH PROGRAM – PhD in PHYSICS
(2024–2025 Batch and onwards)

Course code	Name of the course	Instruction hours / week	credits	Maximum Marks (100)
24RPHY101	Paper I: Research Methodology and Pedagogy	4	4	100
24RPHY201	Paper II: Research Publication Ethics	4	4	100
24RPHY301	Paper III: Physics of Solar Thermal and Photovoltaic Application	4	4	100
24RPHY302	Paper III: Magnetic and Thermo Electric Properties of Materials			
24RPHY303	Paper III:Thin Film Physics			
24RPHY304	Paper III:Crystal Growth			
24RPHY305	Paper III:Solid state Ionics			
24RPHY306	Paper III:Concepts of Nano physics and Nanotechnology			
24RPHY307	Paper III:Laser Physics			
24RPHY308	Paper III:Fluorescence Spectroscopy and Structural Analyze			
24RPHY309	Paper III:Characterization Techniques in Materials Science			
Program Total				

24RPHY101

PAPER – I: RESEARCH METHODOLOGY AND PEDAGOGY

COURSE OBJECTIVES (CO):

- To demonstrate an understanding of the core ideas and principles of research.
- To understand the general definition of research design and the process of designing a research study.
- To adopt ethical issues in research, including those that arise in using quantitative and qualitative research.
- To differentiate between alternative research methodologies and strategies.

COURSE OUTCOMES (COs):

Upon completion of this course, the student will be able to:

COs	Course Outcomes	Blooms Level
CO1	Analyze and present data, both quantitatively and qualitatively	Analyze
CO2	Disseminate research in a way that is consistent with professional practice and research ethics	Apply
CO3	Analyze and evaluate new ideas, and to challenge current assumptions.	Analyze
CO4	Compare and contrast different research methods and paradigms.	Understand
CO5	Identify and prepare the key elements of a research proposal or report.	Apply

UNIT- I INTRODUCTION

Ethics of Research – Objectives of Research – Historical Background of Physics Research – Research Works of Sir C.V. Raman, S.Chandrasekhar and Venkaraman Ramakrishnan (Nobel prize works only) (Nobel Lectures) – Experimental Research in Physics – Design of the experiment, Apparatus to be used, Results and Interpretation – Theoretical Research in Physics – Theory, Models, Methods to solve the problems, results and Interpretation – Literature Survey on Thesis Writing – Online literature survey – Science Citation Index – Impact factor of a journal – Thesis writing

UNIT- II PROBABILITY DISTRIBUTIONS

Mean, Median peak value, and Standard Deviation – Binomial Distribution – Poisson Distribution – Gaussian or Normal Error Distribution – Modes of distributions.

Error Analyze: Instrumental and Statistical uncertainties – Propagation of errors – Estimation of means and errors – Method of least squares – Statistical fluctuations – Chi square test of a distribution

UNIT -III NUMERICAL INTEGRATION

Trapezoidal and Simpson's $1/3$ rule for single integrals - Error estimates - Trapezoidal and Simpson's rule for double integrals

Interpolation: Two points Gaussian quadrature - Three points Gaussian quadrature - Cubic spline interpolation

Eigen values: Power method - Jacobi method (Only 2×2 and 3×3 matrices)

Simulation techniques: Monte Carlo simulation – Fuzzy logic.

UNIT- IV COMPUTER APPLICATIONS IN PHYSICS RESEARCH

Programming in C: Constants - Variables - Data types - Operators and Expressions - Input/Output Statements - Control statements - Functions - Arrays - One, two, multidimensional array declarations and initialization.

Simple applications using C - Program: Program to integrate tabulated function using Trapezoidal rule - Program to integrate tabulated function using Simpson's 1/3 rule - Program to compute the solution of first order differential equation of the type $y' = f(x,y)$ using RK4 method - Program to compute first order differential equation $y' = f(x,y)$ using Milne's method - Program to compute the interpolation value at a specified value from a set of table points using natural cubic spline interpolation

UNIT- VPEDAGOGICAL METHODS IN HIGHER LEARNING

Historical perspectives: Objectives and role of higher education – Learning and learning hierarchy – Information processing – Learning and outcomes – Motivation.

Education evaluation: A conceptual framework – Methods of evaluation – Self evaluation and student evaluation in higher education – Question banking – Diagnostic testing and remedial teaching.

TEXT BOOKS:

1. E.Balagurusamy - Numerical methods , Tata McGraw Hill Publishing company Limited
2. Nye, J.F. (1985). Physical Properties of Crystals: Their Representation by Tensors and Matrices. Oxford University Press, New York.

REFERENCE BOOKS:

1. P.Kandasamy - Numerical methods, K.Thilgavathy and K.Gunavathi, S.Chand and company limited
2. Numerical Mathematical Analyze by Scarborough J B, Oxford &Ibh, ISBN-10: 9788120417595
3. Bevington Philip, Robinson D. Keith – Data Reduction and Error Analyze for Physical Sciences, Mc Graw Hill Higher Education.

WEBSITES:

1. <https://nptel.ac.in/courses/121/106/121106007/>
2. <https://nptel.ac.in/courses/109/105/109105115/>
3. <https://nptel.ac.in/courses/109/103/109103153/>
4. <https://nptel.ac.in/courses/107/108/107108011/>

24RPHY201

PAPER – II: Research Publication Ethics

COURSE OBJECTIVES (CO):

- To study the basics of ethics and philosophy of science
- To understand research integrity and publication ethics
- To identify research misconduct and predatory publications
- To understand about indexing and citation databases
- To know about open access publications, research metrics and plagiarism tools

COURSE OUTCOMES (COs):

Upon completion of this course, the student will be able to:

COs	Course Outcomes	Blooms Level
CO1	Understand the Philosophy of science and ethics: Students learn the basics of philosophy of science and ethics.	Understand
CO2	To have the Knowledge of research integrity.	Apply
CO3	To understanding of publication ethics.	Understand
CO4	To analyze the problem to identify research misconduct and predatory publications.	Analyze
CO5	To have the knowledge of Indexing and citation databases.	Apply

UNIT-I**Philosophy and Ethics:**

Introduction to Philosophy: Definition, nature and scope, concept, branches - Ethics: Definition, moral philosophy, nature of moral judgments and reactions.

UNIT-II**Scientific Conduct:**

Ethics with respect to science and research – Intellectual honesty and research integrity – Scientific misconduct: Falsification – Fabrication and Plagiarism (FFP) - Redundant publications: duplicate and over lapping publications–salami slicing-Selective reporting and misrepresentation of date.

UNIT-III**Publication Ethics:**

Publication Ethics: Definition, introduction and importance - Best practices / standards setting initiatives and guidelines: COPE, WAME, etc. - Conflicts of interest - Publication Misconduct: definition, concept, problems that lead to unethical behavior and vice versa, type - Violation of publication ethics, authorship and contributorship- Identification of publication misconduct, complaints and appeals-Predatory publishers and journals.

UNIT- IV

Publication Misconduct: Group Discussions: Subject specific ethical issues, FFP, authorship-Conflicts of interest- Complaints and appeals: examples and fraud from India and abroad.

Software tools: Use of plagiarism software like Turnitin, Urkund and other open source software tools.

UNIT-V

Databases and Research Metrics:

Database: Indexing database-Citation database: Web of Science, Scopus, etc Research Metrics: Impact Factor of journal as per Journal Citation Report, SNIP, SJR, IPP, Cite Score- Metrics: h-index, g index, i10 index, altmetrics,

UNIT-VI

Development of e-content & IP:

Integrated Library Management System(ILMS):e-journals–e-books–e-shodhsindu–shodhganga – Database - e-content Development - Learning Management System (LMS) – e-PG-Pathshala– CEC (UG)SWAYAM– MOOCs – NPTEL-NMEICT. IPR: Patent–Copyrights-Trademark– Geographical Indication.

PRACTICE

Open Access Publishing:

Open access publications and initiatives - SHERPA / RoMEO online resource to check publisher copy right & self-archiving policies - Software tool to identify predatory publications developed by SPPU - Journal finder / journal suggestion tools viz. JANE, Elsevier Journal Finder, Springer Journal Suggester, etc.

TEXT BOOKS:

1. P.Oliver,(2003),The student's guide to researchethics, Open University PressM aiden head,Philadelphia.
2. H.Zwart,(2010),TalesofResearchMisconduct,SpringerInternational PublishingAG.

REFERENCE BOOKS:

1. Nick Rushbyetal. WileyHandbook ofLearningTechnology.WileyEducation

WEBSITES:

1. Research Impact,<https://guides.osu.edu/c.php?g=608754&p=4224917>.
2. DianeElkinset al.(2015).E-LearningFundamentals:A PracticalGuide

24RPHY301

PAPER – III : SPECIAL PAPER I :

PHYSICS OF SOLAR THERMAL AND PHOTOVOLTIC APPLICATION**Course Objectives**

- To understand the basic concepts of energies produced from various energy sources, advantages and disadvantages
- To learn the present energy scenario and the need for energy conservation
- To facilitate the students to achieve a clear conceptual understanding of technical and commercial aspects of Solar Power Development and Management.
- To enable the students to develop managerial skills to assess feasibility of alternative approaches and drive strategies regarding Solar Power Development and Management.
- To develop a comprehensive technological understanding in solar PV system components

Course Outcomes (COs)

Upon the completion of the course, Students will be able to

COs	Course Outcomes	Blooms Level
CO1	Describe the environmental aspects of non-conventional energy resources	Understand
CO2	Explain the principles that underlie the ability of various natural phenomena to deliver solar energy	Apply
CO3	Outline the technologies that are used to harness the power of solar energy	Understand
CO4	Gain the knowledge about the energy produced from biomass and biogas.	Analyze
CO5	Describe the use of solar energy and the various components used in the energy production	Apply

UNIT- 1 RADIATION GEOMETRY

Basis earth sun angles - Determination of Solar time - Derived Solar angles - Day length - Solar Radiation measurements - selective surfaces - Heat balance energy lost by radiation, convection and conduction - Physical characteristics of selective surfaces - Anti reflection coatings - Solar reflector materials - production methods of coatings.

UNIT -II FUNDAMENTALS OF HEAT TRANSFER

Transfer of Heat by Conduction: Study heat flow in a slab-steady heat flow in a cylindrical shell- Heat transfer through fins – Transient heat conduction. Thermal Radiation: Basic laws of radiation – Radiant heat transfer between two black bodies- Radiant heat transfer between grey bodies. Convection heat loss Evaluation of convective heat transfer co-efficient –Free convection from vertical planes and cylinders – Forced convection – Heat transfer for fully established flow in tubes.

UNIT-III SOLAR THERMAL SYSTEMS

General description of plate collector – thermal losses and efficiency of FPC –Energy balance equation – Evaluation of overall loss coefficient – Thermal Analyze of flat plate collector and useful heat gained by the fluid performance of solar air heaters – Heating and drying of agricultural products Types of drier in use. Solar concentrators and Receiver geometries – General characteristics of focusing collector systems Evaluation of optical losses – Thermal performance of focusing collectors.

UNIT-IV PHOTOVOLTAICS

Description of the photovoltaic effect – Electrical characteristics calibration and efficiency measurement – silicon solar energy converters – Thermal generation of recombination centers silicon. Role of thin films in solar cells Properties of thin films for solar cells CdSe, CdTe, In P, Ga As, Cd Cu₂, Cu In SnO₂, Cd₂SnO₄ ZnO)- Transport properties of metal films – poly crystalline film silicon solar cells (Photovoltaic characteristics, junction Analyze loss mechanisms) Amorphous silicon solar cells (Structural compositional optical and electrical properties)

UNIT- V ENERGY STORAGE AND SOLAR APPLICATIONS

Types of energy storage Thermal storage Latent heat storage – Electrical storage principle of operation of solar ponds-Non convective solar ponds – Theoretical Analyze of solar pond – solar distillation – solar cooking –solar pumping.

TEXT BOOKS:

1. Charles E. Backus (1976). Solar cells. IEEE Press
2. Garg, H.P. (1982). Treatise on solar energy volume I fundamentals of Solar Energy.
3. Kasturi Lal Chopra and Suhit Ranjan Das (1983). Thin film solar cells.

REFERENCE BOOKS:

1. Rai, G.D. (1996). Solar energy utilization.
2. Rai, G.D. Thermal performances testing of FPC and CPC.

WEBSITES:

1. <https://nptel.ac.in/courses/115/107/115107116/>
2. <https://nptel.ac.in/courses/115/103/115103123/>
3. https://onlinecourses.nptel.ac.in/noc19_ph13/preview
4. https://onlinecourses.nptel.ac.in/noc20_mm05/preview

24RPHY302

PAPER – III: SPECIAL PAPER II:

MAGNETIC AND THERMOELECTRIC PROPERTIES OF MATERIALS**Course objectives (Co):**

- To know about the origins of magnetism, diamagnetism, paramagnetism, crystal field environments, dipolar and exchange interactions, ferromagnetism, antiferromagnetism, magnetic domains, magnetic anisotropy and magnetostriction.
- To give an idea in the experimental techniques in magnetic characterization.
- To study different magnetic materials, include transition metals, their alloys and oxides, rare earths and their oxides, organic and molecular magnets.
- To study the basic concepts of thermoelectric materials and their phenomenon
- To study the interaction of phonon electron scattering in semiconductor materials

Course Outcomes (Cos)

Upon the completion of the course students will able to

COs	Course Outcomes	Blooms Level
CO1	Use the models of magnetization and remagnetization in various cases;	Understand
CO2	Calculate the contribution of magnetostatic energy, elastic energy and anisotropy in magnetic properties;	Apply
CO3	Solve theoretical and applied problems connected with professional activities	Understand
CO4	Explain transport mechanisms (electronic and thermal transport), guiding the thermal/thermoelectric performance of any materials system	Analyze
CO5	Describe the different techniques for thermal/thermoelectric characterization of the samples with varying geometries.	Apply

UNIT -I

Introduction -Classification of Magnetic Materials-Factors Affecting Permeability and Hysteresis Loss- Magneto statics - The magnetic dipole moment - Magnetic fields - Maxwell's equations - Magnetic field calculations - Magneto static energy and forces - Magnetism of electrons - Orbital and spin moments - Magnetic field effects-Theory of electronic magnetism -Magnetism of electrons in solids.

UNIT-II

Magnetism of localized electrons on the atom - The hydrogenic atom and angular momentum - The many-electron atom - Paramagnetism - Ions in solids; crystal-field interactions - Ferromagnetism and exchange - Mean field theory - Exchange interactions - Band magnetism -Collective excitations Anisotropy - Ferromagnetic phenomena - Anti ferromagnetism and othermagnetic order -Molecular field theory of anti-ferromagnetism - Ferrimagnets -Frustration - Amorphous magnets -Spin glasses - Magnetic models.

UNIT-III

Experimental methods - Materials growth - Magnetic fields - Atomic-scale magnetism - Domain-scale measurements - Bulk magnetization measurements -Excitations -Numerical method - Magnetic materials - Iron group metals and alloys -Rare-earth metals and intermetallic compounds - Interstitial compounds - Oxides with ferromagnetic interactions - Oxides with antiferromagnetic interactions - Miscellaneous materials.

UNIT –IV

The Thermoelectric and Related Effects -Introduction - Relations Between the Thermoelectric Coefficients Effects in a Magnetic Field - The Transport Effects - Thermoelectric Refrigerators and Heat Pumps - Thermoelectric Generators - Multi-Stage Devices –Application of the Thermomagnetic Effects - Optimisation and Selection of Semiconductor Thermoelements - Power Factor – The Materials Parameter -Mobility and Effective Mass - The Lattice Thermal Conductivity in Pure Crystals. The Effect of Temperature -The Importance of the Energy Gap

UNIT –V

Thermoelectric Properties of Metals and Semiconductors - Transport by Electrons Metals and Semiconductors - Bipolar Effects - Phonon Conduction - Phonon Drag -Minimising the

Thermal Conductivity - Semiconductor Solid Solutions -Phonon Scattering by Point Defects - Boundary Scattering -Scattering of Electrons and Phonons - Fine-Grained Material with Large Unit Cells - Phonon-Glass Electron-Crystal-Applications of thermo electric materials.

TEXT BOOKS:

1. Introduction to Magnetic Materials, B. D. Cullity, C. D. Graham, John Wiley and sons, Inc, Publications,2009
2. Magnetism and Magnetic Materials, [J. M. D. Coey](#), Cambridge University Press, 2012
3. The Quantum Theory of Magnetism, Second Edition, Norberto Majis, World Scientific Publishing Co. Pte. Ltd. 2007

REFERENCE BOOKS:

1. Magnetism in Condensed Matter, Stephen Blundell, Oxford University Press, 2011
2. CRC Handbook of Thermoelectrics, Edited by D.M. Rowe, Ph.D., D.SC, CRC Press, Boca Raton London New York, 2004
3. Introduction to Thermoelectricity H. Julian Goldsmid, Springer Series in materials science *Edited by* R. Hull R. M. Osgood, Jr. J. ParisiH.Warlimont, 2009
4. Novel Thermoelectric Materials and Device Design Concepts, Editors: Skipidarov, Sergey, Nikitin, Mikhail (Eds.), Springer International Publishing,2019

WEBSITES:

1. <https://nptel.ac.in/content/storage2/courses/122101002/downloads/lec-21.pdf>
2. https://nptel.ac.in/content/storage2/courses/112108150/pdf/Web_Pages/WEBPM16.pdf
3. <https://www.nature.com/articles/nnano.2016.182>

Course objectives (Co):

- To Introduce physical concepts and mathematical tools used to describe surfaces, interfaces and thin films.
- To develop an intuition for surface and thin film physical principles through plotting of functions using Maple
- To relate the mathematical results to practical applications and experiments in thin film techniques.
- To Develop an appreciation of the mathematical basis for experimental techniques for deposition and Analyze of thin films
- To Understand physical phenomena that can be exploited for the deposition of thin films

Course Outcomes (Cos)

Upon the completion of the course students will able to

COs	Course Outcomes	Blooms Level
CO1	Evaluate and use models for nucleating and growth of thin films	Analyze
CO2	Examine the relation between deposition technique, film structure, and film properties, discuss typical thin film applications	Apply
CO3	Select proper deposition techniques for various applications	Understand
CO4	Understand the basic concepts about the thin film technology	Apply
CO5	To gain experience in handling high vacuum equipment and using thin film growth techniques which enables them to work at production units related to optical, mechanical, electronic coatings etc.	Apply

UNIT -I PREPARATION OF THIN FILMS

Spray pyrolytic process – characteristic feature of the spray pyrolytic process – ion plating – Vacuum evaporation – Evaporation theory – The construction and use of vapour sources – sputtering Methods of sputtering – Reactive sputtering – RF sputtering - DC planar magnetron sputtering.

UNIT- II THICKNESS MEASUREMENT AND NUCLEATION AND GROWTH IN THIN FILM

Thickness measurement: electrical methods – optical interference methods – multiple beam interferometry – Fizeau – FECO methods – Quartz crystal thickness monitor. Theories of thin film nucleation – Four stages of film growth incorporation of defects during growth.

UNIT- III ELECTRICAL PROPERTIES OF METALLIC THIN FILMS

Sources of resistivity in metallic conductors – sheet resistance - Temperature coefficient of resistance (TCR) – influence of thickness on resistivity – Hall effect and magneto resistance – Annealing – Agglomeration and oxidation.

UNIT- IV TRANSPORT PROPERTIES OF SEMICONDUCTING AND INSULATING FILMS

Semiconducting films; Theoretical considerations - Experimental results – Photoconduction – Field effect thin films – transistors, Insulation films Dielectric properties – dielectric losses – Ohmic contacts – Metal – Insulator and Metal – metal contacts – DC and AC conduction mechanism .

UNIT- V OPTICAL PROPERTIES OF THIN FILMS AND THIN FILMS SOLAR CELLS

Thin films optics – Theory – Optical constants of thin films – Experimental techniques – Multilayer optical system – interference filters – Antireflection coating, thin films solar cells: Role, Progress, and production of thin solar cells – Photovoltaic parameter, thin film silicon (Poly crystalline) solar cells : current status of bulk silicon solar cells – Fabrication technology – Photo voltaic performance: Emerging solar cells: GaAs and CuInSe_2 .

TEXT BOOKS:

1. Anderson, J.C. The use of thin films in physical investigation.
2. Berry, Koil and Harris. Thin films technology.
3. Chopra, K.L. Thin film Phenomena.

REFERENCE BOOKS:

1. Chopra, K.L. and Das, S.R. Thin films solar cells.
2. George Hass and others (Ed). Physics of thin films, vol. 12.
3. Holland, L. Vacuum deposition of thin films.
4. Maissel, L.I. and Clang, R. Hand book of Thin films Technology.
5. Vilsan, J.L. Thin films processes.

WEBSITES:

1. <https://nptel.ac.in/content/storage2/courses/112108092/module2/lec08.pdf>
2. <http://www.infocobuild.com/education/audio-video-courses/materials-science/FundamentalsOfMaterialProcessing2-IIT-Kanpur/lecture-37.html>
3. <https://www.youtube.com/watch?v=H2h0tz5KfPw>
4. <https://www.youtube.com/watch?v=p0XxWT2QdEk>

24RPHY304**PAPER – III: SPECIAL PAPER IV: CRYSTAL GROWTH****Course Objectives (Co):**

- To strengthen the students with crystallographic and crystal growth techniques
- To provide the general characteristics of crystals, methods of preparation etc.
- Various thin films deposition techniques and thin film characterization techniques are also covered in the course.
- To give an idea about historical importance of crystals, methods of preparation and characterization of crystals etc.
- To explore the knowledge in fundamentals of materials syntheses, crystal growth techniques, zone refining, properties etc.,

Course Outcomes (COs)

Upon the completion of the course students will able to

COs	Course Outcomes	Blooms Level
CO1	Learn about the crystal growth mechanisms and techniques.	Understand
CO2	Understand different crystals having a lot applications in electronics, energetics etc.	Apply
CO3	Acquire the theoretical concept behind electrical and thermal properties of metals	Understand
CO4	Gain the knowledge about Semiconductor Crystals and their properties	Apply
CO5	Gain the knowledge about phonons and its importance in thermal physics	Apply

UNIT- I FUNDAMENTALS OF CRYSTAL GROWTH

Importance of crystal growth – Classification of crystal growth methods – Basic steps: Generation, transport and adsorption of growth reactants – Nucleation: Kinds of nucleation – Classical theory of nucleation: Gibbs Thomson equations for vapour and solution – Kinetic theory of nucleation – Becker and Doring concept on nucleation rate – Energy of formation of a spherical nucleus – Statistical theory on nucleation: Equilibrium concentration of critical nuclei, Free energy of formation.

UNIT- II THEORIES OF CRYSTAL GROWTH

An introductory note to Surface energy theory, Diffusion theory and Adsorption layer theory – Concepts of Volmer theory, Bravais theory, Kossel theory and Stranski's treatment – Two-dimensional nucleation theory: Free energy of formation, Possible shapes and Rate of nucleation – Mononuclear, Polynuclear and Birth and Spread models – Modified Birth and Spread model – Crystal growth by mass transfer processes: Burton, Cabrera and Frank (BCF) bulk diffusion model, Surface diffusion growth theory.

UNIT-III MELT GROWTHTECHNIQUES.

Basics of melt growth – Heat and mass transfer – Conservative growth processes: Bridgman-Stockbarger method – Czochralski pulling method – Kyropoulos method – Nonconservative processes: Zone-refining – Vertical and horizontal float zone methods – Skull melting method – Vernueil flame fusion method.

UNIT -IV SOLUTION GROWTHTECHNIQUES.

Growth from low temperature solutions: Selection of solvents and solubility – Meir's solubility diagram – Saturation and supersaturation – Metastable zone width – Growth by restricted evaporation of solvent, slow cooling of solution and temperature gradient methods– Crystal growth in Gel media: Chemical reaction and solubility reduction methods – Growth from high temperature solutions: Flux growth Principles of flux method – Choice of flux – Growth by slow evaporation and slow cooling methods – Hydrothermal growth method.

UNIT -V VAPOUR GROWTHTECHNIQUES

Basic principles – Physical Vapour Deposition (PVD): Vapour phase crystallization in a closed system – Gas flow crystallization – Chemical Vapour Deposition (CVD): Advantageous and disadvantageous – Growth by chemical vapour transport reaction: Transporting agents, Sealed capsule method, Open flow systems – Temperature variation method: Stationary temperature profile, Linearly time varying temperature profile and Oscillatory temperature profile.

TEXT BOOKS:

1. Brice, J.C. (1986). Crystal Growth Processes. John Wiley and Sons, New York.
2. Mullin, J.W. (2004), Crystallization. Elsevier Butterworth-Heinemann, London.
3. Pamplin, B.R. (1975). Crystal Growth. Pergamon Press, Oxford.

REFERENCE BOOKS:

1. Sunagawa Ichiro. (2005). Crystals: Growth, Morphology and Perfection. Cambridge University Press, Cambridge.
2. Vere, A.W. (1987). Crystal Growth: Principles and Progress. Plenum Press, New York.

WEBSITES:

1. https://nptel.ac.in/content/storage2/courses/103104045/pdf_version/lecture19.pdf
2. https://www.youtube.com/watch?v=G76H7A6_iyo
3. <https://www.youtube.com/watch?v=8GsyipwtCIA>
4. <https://www.youtube.com/watch?v=db5nZCipJh8>

Course Objectives (Co):

- To provide an introduction to the concepts underlying solid state Ionics
- To illustrate the wide range of materials and physical properties that currently available for ionic conductors
- To introduce the superionic conductors and their applications
- To establish the ionic conductors for energy applications
- To introduce the different mechanism of electrochemical energy storage materials and their applications

Course Outcomes (COs)

Upon the completion of the course students will able to

COs	Course Outcomes	Blooms Level
CO1	Calculate point defect concentrations using formation energies, develop Brouwer diagrams, describe several means of tailoring point defect concentrations through independent variables, and apply equilibrium thermodynamics to the case of defective solids	Understand
CO2	Understand point defect reactions in Kroger-Vink notation to describe defect processes, and apply a non-equilibrium thermodynamics and chemical kinetics framework to describe defect reactions and kinetic behavior	Apply
CO3	Describe operation of various solid state ionics applications (including open circuit cells, cells using current, and cells generating current)	Understand
CO4	Learn the superionic conductors and their real-life applications.	Apply
CO5	Learn synthesis design and planning, different processing technoques and their chemical- physical fundamentals as well as basic method of characterisation of solids.	Apply

UNIT- I Introduction

Crystalline solids – space lattice – the basis and crystal structure; crystal translational vectors, symmetry operation primitive lattice cell and unit cell symmetry elements, Fundamental type of lattice, atomic packing, atomic radius, lattice constants and density, crystal structure other cubic structure – type of bonding – Ionic bonding – Energy of formation of NaCl molecules, Madelung constants – potential energy of diagram of ionic molecules – calculation of repulsive exponent – Born Haber cycle characteristics of ionic bond.

UNIT- II Transport Properties of Ionic Conductors

Ionic conductivity – Normal and super ionic conductors – Mass transport in crystals – Diffusion – Atomic diffusion theory – Experimental determination of the diffusion constant – Ionic conduction – Experimental results – for ionic conduction – The Einstein relation – Dielectric loss in ionic crystals – Electronic conduction in ionic crystals – Excess conductors – Deficit conductors –

Amphoteric semiconductor.

UNIT- III Phenomenological Models

Huberman's Theory – Riesstrassler Toom's Theory – Weleh and Diene Theory – Lattice Gas theory – Free ion model – Domain Model – Rica and Roth Theory – The Path Probability Method – The static variables – the Path variables – The path Probability – Stationary state condition – Classification of Superionic solids – Crystalline and Amorphous – Glasses – Dispersed solid Electrolytes – polymers – Ion exchange resins – biological basis resins – Classification over conducting ion species – mode and mechanism of conduction in each case and their corresponding criteria to be superionic conductors.

UNIT IV Experimental Techniques and Methods

Structural characterization – XRD surface Analyze, EXAFS, IPS and Quasi neutron scattering – Thermo dynamical characterization – Differential scanning calorimetry, Differential Thermal Analyze, Thermo Gravimetric Analyze and Thermo electric power – Ion transport properties – Electrical conductivity – Two probe method – four probe method – Immitance spectroscopy – Dynamical conductivity – state conductivity – polarisation characteristic – determination of small electronic transport numbers.

UNIT- V Electrochemical Techniques and Applications

Fundamentals of electrochemistry, Linear Sweep Voltammetry, Cyclic Voltammetry, Chronoamperometry, Linear polarization, Electrochemical Impedance spectroscopy. Batteries: Primary and secondary batteries, Li-ion batteries, Supercapacitors: Electric double layer capacitor, Pseudocapacitor, Fuel Cells: Solid oxide Fuel cells, Direct Methanol Fuel Cells, Proton Exchange Membrane Fuel cells, Sensors: Oxygen sensors and electrochemical sensors, Electrochromic displays.

TEXT BOOKS:

1. Superionic solid – Principles and applications (Ed. S.Chandra) North Holland 1981.
2. Solid state ionics (Eds. T Kudo and Fueki) VCH Publishers, Kodansha 1990.
3. Lectures on solid state physics (Eds. G Bush and H Schade), international series on Natural Philosophy Vol. 79 Pergamon, press 1976.

REFERENCE BOOKS:

1. "Solid Electrolytes" (Eds. S Geller) Springer Verlag New York 1977.
2. Impedance Spectroscopy Theory, Experiment, and Applications, (Eds) EvgenijBarsoukov and J. Ross Macdonald, Wiley interscience (2005).
3. Physics of Electrolytes – Transport Processes solid Electrolytes and in Electrodes (Eds. J Hladik) Academic press, New york 1972.
4. Fundamentals of Electrochemistry, 2nd Edition, V.S.Bagotsky, Wiley Interscience. (2006).
5. Electrochemical Methods: Fundamental and Application, Allen J.Bard Wiley and Sons

Publications (2001).

WEBSITES:

1. https://nptel.ac.in/content/storage2/courses/113104005/lecture_pdf/module3.pdf
2. https://nptel.ac.in/content/storage2/courses/112108150/pdf/PPTs/MTS_14_m.pdf
3. <https://nptel.ac.in/content/storage2/courses/122101002/downloads/lec-8.pdf>
4. <https://nptel.ac.in/content/storage2/courses/103108100/module6/module6.pdf>

24RPHY306**PAPER – III: SPECIAL PAPER VI :
CONCEPTS OF NANOPHYSICS AND NANOTECHNOLOGY****Course Objectives (Co):**

- To foundational knowledge of the Nanoscience and related fields.
- To make the students acquire an understanding the Nanoscience and Applications
- To help them understand in broad outline of Nanoscience and Nanotechnology.
- To familiarize with the on-going merge of the top-down approach of microelectronics and micromechanics with the bottom-up approach of chemistry/biochemistry.
- To demonstrate the potential of nanoscience and industrial applications of nanotechnology.

Course Outcomes (COs):

Upon the completion of the course students will able to

COs	Course Outcomes	Blooms Level
CO1	Understand the physical basis of new phenomena that appear when the linear dimension of an object or device shrinks below a micrometer	Understand
CO2	become familiar with the whole concept of nanoscale science and technology and be able to apply their knowledge for understanding further developments in this rapidly emerging area.	Apply
CO3	Acquire the basic knowledge of the physical phenomena, theoretical concepts and experimental techniques behind the recent vastly improved ability to observe, fabricate and manipulate individual structures on the nanometer scale.	Understand
CO4	Apply their learned knowledge to develop Nanomaterial's	Apply
CO5	Understand the synthesis of nanomaterials and their application and the impact of nanomaterials on environment	Apply

UNIT- I INTRODUCTION TO NANOTECHNOLOGY

Defining nanotechnology, Historical development – Beyond Moore's law, Comparison of bulk and nano materials – change in band gap and large surface to volume ratio, Classification of nanostructured materials – one, two and three-dimensional confinement, quantum dots, quantum wires and quantum wells, scope of applications.

UNIT- II SYNTHESIS AND CHARACTERIZATION

Classification of fabrication methods – Top to bottom approach – Ball milling, etching etc bottom to top approach – Physical and chemical methods – Molecular Beam Epitaxy, optical and electron beam lithography, Ion implantation, sputtering, thermal evaporation, pulsed laser deposition, chemical vapor deposition, controlled precipitation, sol gel methods. Grain size determination – XRD (Debye Scherer equation), TEM, AFM, STM and Light scattering techniques. Composition Analyze – ICP – AES, EDAX, SIMS.

UNIT- III OPTICAL AND VIBRATIONAL PROPERTIES OF NANOPARTICLES

Basic concepts – Band structure of solids, excitons, effective mass, reciprocal lattice, Brillouin zone, phonons etc. Size and dimensionality effects – Bulk to nano transition –Density of states, potential well - quantum confinement effect – weak and strong confinement regime. Blue shift of band gap - Effective mass approximation (Rigorous mathematical treatment not necessary). Phonon confinement effect and presence of surface modes. Characterization tools - UV – Visible absorption and Photoluminescence techniques, Raman and IR spectroscopy

UNIT -IV CARBON NANOSTRUCTURES

Carbon nanostructures – carbon molecules – carbon clusters. Fullerene - structure of C₆₀ and its crystal – larger and smaller fullerenes – other bucky balls. Carbon nanotubes – fabrication – structure – electrical properties – vibrational properties – mechanical properties. Applications of carbon nanotubes – Field emission and shielding – computers – Fuel cells – Chemical sensors – Catalysis – Mechanical reinforcement.

UNIT- V NANOMACHINES AND NANODEVICES

Extension of conventional devices by nanotechniques – Bipolar and MOS transistors – structure and technology, electrical characteristics, limitations, low temperature behavior. Microelectromechanical systems (MEMSs), Nanoelectromechanical systems (NEMSs), Resonant Tunneling Diode, Quantum Cascade lasers, Single Electron Transistors – Operating principles and applications.

TEXT BOOKS:

1. Mick Wilson, Kamali Kannangara, Geoff Smith, Michelle Simmons and Burkhard Raguse “Nanotechnology”, Overseas Press New Delhi 2005

REFERENCE BOOKS:

1. W. R. Fahrner (Ed.) “Nanotechnology and Nanoelectronics”, Springer 2006.
2. Charles P Poole Jr and Frank J Owens “Introduction to Nanotechnology”, Wiley student edition 2003.

WEBSITES:

1. <https://nptel.ac.in/courses/118/104/118104008/>
2. <https://nptel.ac.in/courses/115/101/115101007/>
3. <https://nptel.ac.in/courses/113/106/113106093/>

Course Objectives (Co):

- Laser is a versatile tool with applications in almost all fields from medical to astronomy, communications, welding, cutting etc.
- This paper explains the characteristics of lasers, different types of lasers and their construction to apply for industrial use. Applications of lasers in different fields are also explained.
- To provide up-to-date guidance of modern types of lasers and will give sufficient theoretical and, importantly, practical knowledge for designing and building actual lasers.
- To give exposure to students about the characteristics of different lasers, their fabrication techniques, applications etc.
- To make the student understand the principles of Lasers

Course Outcomes (COs):

Upon the completion of the course students will able to

COs	Course Outcomes	Blooms Level
CO1	Acquire fundamentals and principles of Laser action and understand the basic concepts of different types of lasers	Understand
CO2	Learn the basics & different parameters required to fabricate the lasers and their advantages and disadvantages in various fields.	Apply
CO3	Explain operational principles and construction of lasers	Understand
CO4	Describe modes in multimode fibers and mode field parameter in single-mode fibers	Apply
CO5	Relate the laser operation principles to atom and molecular physics, solid state physics, quantum mechanics and physical optics.	Apply

UNIT -I

Radiative transitions and emission line widths. Radiative decay of excited states, homogeneous and inhomogeneous broadenings. Absorption, spontaneous and stimulated emissions. Einstein's A and B Coefficients. Absorption and gain of homogeneously broadened radiative transitions, gain coefficient and stimulated emission cross section for homogeneous and inhomogeneous broadening.

UNIT -II

Necessary and sufficient conditions for laser action (population inversion and saturation intensity), threshold requirements for laser with and without cavity, laser amplifiers, rate equations for three and four level systems, pumping mechanisms. Laser cavity modes- longitudinal and transverse modes in rectangular cavity. FP cavity modes, Spectral and spatial hole burning, stability of laser resonator and stability diagram, unstable and ring resonators.

UNIT III

Q-switching and Mode locking, active and passive techniques, generation of giant pulses and pico second optical pulses, Properties of laser beam and techniques to characterize laser beam.

UNIT -IV

Scattering: Scattering cross-section – Scattering amplitude – Partial waves – Scattering by a central potential: partial wave Analyze – Significant number of partial waves – Scattering by an attractive square-well potential – Briet-Wigner formula – Scattering length – Expression for phase shift – Integral equation – The Born approximation – Scattering by screened coulomb potential – Validity of Born approximation - Laboratory and center of mass co-ordinate systems.

UNIT -V

Introduction - Driving problems in biomedical imaging - Sources of imaging data: acquisition and noise - Elementary image processing - Grenander's Pattern Theory, Biomedical image Analyze using MATLAB – Image registration – unaided and Interactive – Segmentation – Edge detection – Real time imaging applications.

TEXT BOOKS:

1. Laser Fundamentals - W T Silfvast, Cambridge University Press (1996)(Text)
2. Laser Electronics - J T Vardeyan. PHI, 2nd Ed (1989)
3. Lasers-Theory and Applications- Ghatak and Thyagarajan, McMillan (2002) (Text)

REFERENCE BOOKS:

1. Principles of lasers - Svelto, Plenum Press (1948)
2. Solidstate laser engineering - Koechner, Springer Verlag (1993)
3. Laser Physics- Tarasov. Mir Publishers (1985)
4. John.L.Semmlow, Biomedical signal and Biomedical Image Processing – MATLAB based applications, Marcel Dekker Inc., 2004.
5. Rangaraj M. Rangayyan, Biomedical Image Analyze, CRC press.
9. <https://nptel.ac.in/courses/104/104/104104085/>

WEBSITES:

1. http://www.cdeep.iitb.ac.in/webpage_data/nptel/Electrical%20&%20Comm%20Engg/Optical%20Communication-backup/Course_home-M7.html
2. <https://www.digimat.in/nptel/courses/video/104104085/L01.html>
3. [fundamentals-of-photonics-quantum-electronics-spring-2006/lecture-notes/chapter7.pdf](#)
4. <https://ocw.mit.edu/courses/electrical-engineering-and-computer-science/6-974->

FLUORESCENCE SPECTROSCOPY AND STRUCTURAL ANALYSICS**Course Objectives (Co):**

- To know the modern optical spectroscopic and imaging techniques and their applications to biology and chemistry.
- To get the knowledge an introduction to fundamental concepts of light-matter interaction, lasers and laser systems, detectors and other relevant aspects of instrumentation necessary for spectroscopy and imaging.
- To discuss various modern surface spectroscopic techniques and examples from classic and contemporary literature.
- To get an in-depth introduction to the principles of fluorescence spectroscopy and its applications to the Life Sciences.
- To gain the knowledge in the advanced X-ray diffraction techniques for opto-electronic materials characterization.

Course Outcomes (COs):

Upon the completion of the course students will able to

COs	Course Outcomes	Blooms Level
CO1	Explain the fundamental physical mechanisms involved in the generation of fluorescence light.	Understand
CO2	Mention the most important fluorescence techniques in the biomedical research field, and explain what type of questions these techniques can address.	Apply
CO3	Describe the physical principles of these fluorescence techniques	Understand
CO4	Based on knowledge on these techniques and their physical principles, describe and motivate what the factors are that limit their performance, and how the obtained measurements data are evaluated.	Apply
CO5	explain how interactions between biomolecules and electromagnetic radiation and environmental effects can generate changes in the measured fluorescence parameters, and how these changes can be exploited for monitoring of biomolecules and their interactions.	Apply

UNIT -I Solvent and Environmental Effects on Fluorescence spectra

Stokes' shifts and solvent relaxation, general and specific solvent effects, other mechanisms for spectral shifts. Lippert equation, Derivation of Lippert equation, Applications of Lippert equation, Specific solvent effects. Temperature effects, Additional factors that affects the emission spectra - locally excited and internal charge transfer states, excited state intramolecular proton transfer,

effects of viscosity, probe-probe interaction and effect of solvent mixtures.

UNIT - II Fluorescence Quenching

Introduction, quenchers of fluorescence, Theory of colloidal quenching, Derivation of Stern-Volmer equation, Interpretation of bimolecular quenching constants, theory of static quenching, Comparison between static and dynamic quenching. Combined dynamic and static quenching with examples. Deviation from the Stern-Volmer equation - Quenching sphere of action. Derivation of the quenching sphere of action, Origin of the Smoluchowski equation.

Mechanisms and Dynamics of Fluorescence Quenching

Introduction, comparison of quenching and resonance energy transfer, distance dependence of resonance energy transfer and quenching, encounter complexes and quenching efficiency, mechanisms of quenching: Intersystem crossing or heavy atom effect, electron exchange, photoinduced electron transfer. Transient effects in quenching.

Fluorescence Sensing

Optical Clinical Chemistry and spectral observable, spectral observable for fluorescence sensing, Mechanism of sensing, sensing collisional quenching - oxygen sensing, chloride sensors, energy transfer sensing - pH and pCO₂ sensing by energy transfer, glucose sensing by energy transfer, ion sensing by energy transfer, theory of energy transfer sensing.

UNIT-III X-RAY CRYSTALLOGRAPHY

Crystal and Symmetry: Growth of single crystals, different methods, Optical properties, ferro electric, piezoelectric, thermal properties of crystal, Crystal system- Bravais lattices- point group and space group, symmetry elements.

Quasi crystals: definition, preparation, symmetry orientation order in quasicrystals, Quasi-periodic space tiling procedure. Macro molecules: definition, examples of macro molecules or Biomolecules-symmetry.

X-rays: Production, white radiation characteristics, radiation - absorption edge, filters - absorption by crystals.

UNIT-IV DIFFRACTION OF X-RAYS

Direct and reciprocal lattice, Ewald's sphere and Bragg's law, Spacing formula, Transformation equations, Interpretation of rotation photograph. Scattering of X-rays by a distribution of electron, structure factor, calculation of electron density function, Fourier synthesis, the crystal symmetry and x-ray diffraction pattern, Friedel's law and its break down. Electron and neutron diffraction, comparison with X-ray diffraction, significance of electron and neutron diffraction, characterization of quasi crystalline sample using electron diffraction. The Laue method, The Powder method, rotation and Weissenberg methods, The Burger precession method.

UNIT-V INTENSITY DATA COLLECTION, STRUCTURE SOLUTION AND REFINEMENT

The single crystal diffractometer method, intensity data collection, corrections to intensity data- Lorentz, polarization, spot shape and absorption effects, primary and secondary extinction effects, absolute scaling and temperature factors.

Fourier techniques, Phase problem, Patterson function and its significance, Heavy atom methods, Isomorphous replacement method, anomalous scattering method, direct methods.

Cyclic Fourier refinement, the difference Fourier refinement, correction for series termination effects, temperature correction, Least squares refinement. Derived results- bond lengths, bond angles, standard deviations in bond lengths and angles, comparison and averaging of bond lengths and angles, least square planes, absolute configuration and thermal motion.

TEXT BOOKS:

1. Fundamentals of Photochemistry, Rohtagi - Mukherjee K K, Wiley Eastern Ltd., 1992.
2. Principles of Fluorescence Spectroscopy, Joseph R Lakowicz, Plenum Press, New York, 1986
3. Photophysics of Aromatic Molecules, Birks J B, Wiley - Interscience, London 1970.

REFERENCE BOOKS:

1. Azaroff. L.V.: Introduction to Solids, McGraw-Hill, New York, 1960.
2. Phillips. F.C. :Introduction to Crystallography, Longmans, London, 1966.
3. Cullity. B. D.: Elements of X-ray crystallography, prentice hall, 2001.
4. Ponerger. J. J.: X-ray Crystallography, John Wiley, New York, 1942.
5. Burger. M. J.: Crystal Structure Analyze, John Wiley, New York, 1960.
6. 9 Stout. H & Jensen. L. H.: X-ray Structure determination, McGraw Hill, London, 1973.
7. Duncan Mc Kie&Christins Mc Kie: Crystalline Solids, Nelson, London, 1973.
8. Azaroff. L.V. Elements of X-ray crystallography, McGraw-Hill , New York, 1968.
9. Woolfson, M. M.: X-ray Crystallography, Cambridge University Press, 1978.
10. Glusker, J. P. & True blood. K.N.: Crystal Structure Analyze, Oxford Univ. Press, 1985.
11. Bacon. G. E.: Neutron Diffraction, Oxford Univ. Press, 1962.
12. Methods of Experimental Physics, Vol. 6: Part A, Associate Press.
13. Janot. C, Quasicrystals, Oxford Science Publications, Clarendon press, Oxford, 1992.

WEBSITES:

1. <https://nptel.ac.in/noc/courses/noc17/SEM2/noc17-cy01/>
2. <https://nptel.ac.in/courses/104/104/104104084/>
3. <http://web.iitd.ac.in/~sdeep/Fluorescence.pdf>

Course Objectives (Co):

- To learn how to operate a number of important materials processing and characterization instruments as well as to analyze and interpret the resultant data.
- To learn the strengths and weaknesses of different materials processing and characterization techniques.
- To gain a better understanding of the important processing-structure-property relationships.
- To provide concepts on the several materials characterization techniques at the morphological, structural and chemical level, the acquisition of skills in the use and selection of advanced experimental techniques for characterization of materials.
- To solving problems in materials science and engineering. Several characterization techniques are discussed, from the most conventional to the most recent.

Course Outcomes (COs):

Upon the completion of the course students will able to

COs	Course Outcomes	Blooms Level
CO1	Know the proper selection of material based on the performance of the system under study and development of new materials.	Understand
CO2	Understand samples preparation and the applicability of these techniques in order to provide the essential groundwork for select and ranking them	Apply
CO3	Know the advanced techniques for materials characterization, particularly of the most widely used materials as thin films, nano materials and advanced materials.	Understand
CO4	Know the main techniques in the study include characterization methods based on microscopy, micro Analyze and diffraction techniques, and surface and spectroscopy Analyze.	Apply
CO5	Understand the solving problems in materials science and engineering.	Apply

UNIT -1 METHODS OF MATERIALS BULK SYNTHESIS

Solid state reaction - ceramic technique - microwave synthesis - sol-gel method - wet-chemical methods - Hydrothermal method.

Growth of Single Crystals - Introduction to Methods of Growth of Crystals –BCF theory- Czochralski Method - Bridgman, Zone Melting and Zone Refining Methods.

UNIT-2 PREPARATION OF THIN FILMS

Types of thin Film Growth process - Spin coating - vacuum evaporation - sputtering - Pulsed laser deposition - Vapor Methods – CVD – PVD - Fundamental aspects of Epitaxial Growth methods.

UNIT – 3 STRUCTURAL CHARACTERIZATIONS AND IMAGING TECHNIQUES

X-ray diffraction ((XRD) - Electron and neutron diffraction - elementary ideas of photoelectron spectroscopy (PES) - Basic principle of atomic resolution electron microscopy - Scanning and Transmission electron microscopy (SEM, TEM) - Scanning tunneling and atomic force microscopy (STM, AFM) techniques.

UNIT-4 OPTICAL CHARACTERIZATIONS AND SPECTROSCOPIC TECHNIQUES

Ultraviolet / visible (UV/Vis) absorption spectroscopy - Raman and Infrared Spectroscopy - Fluorescence spectroscopy - Elementary idea of laser based non-linear techniques. Room temperature as well as low temperature Photoluminescence - Cathode Luminescence - Mössbauer spectroscopy - Impedance spectroscopy

UNIT-5 PHYSICAL PROPERTY MEASUREMENTS

Intensive and extensive properties - Physical property measurements (DSC, DTA, TGA,) - Transport properties (R-T) – Photoconductivity study (C-V, I-V) - Low conductivity measurement (Dielectric Spectroscopy) - P-E loops for ferro electrics - magnetic properties of bulk and nano phases of material (VSM & SQUID).

TEXT BOOKS:

1. H.H. Willard, L.L. Merritt, J.A. Dean and F.A. Settle, Instrumental Methods of Analyze, 6th Ed., C.B.S. Publishers, New Delhi, 1991.
2. G.A. Higgerson, Experiments in Materials Technology, Affiliated East-West Press, 1973.
3. L.C. Azzarof, Elements of X-ray Crystallography, McGraw-Hill, New York, 1968.

REFERENCE BOOKS:

1. M.V. Heimendahl, Electron Microscopy of Materials-An Introduction, Academic Press, 1980.
2. Elton N. Kaufmann, Characterization of Materials volumes 1 and 2, John Wiley & Sons, Inc., Hoboken, New Jersey, 2003.
3. L. E.Murr. Electron and Ion microscopy and MicroAnalyze principles and Applications.Marcel Dekker Inc., New York, 1991.
4. V.Raghavan, Materials Science for Engineering, Prentice Hall of India Pvt Ltd, 2006.

5. Meissel. L.T and R. Glang., 2000 Handbook of thin film technology, Tata McGraw Hill, New Delhi.

WEBSITES:

1. <https://nptel.ac.in/courses/113/102/113102080/>
2. <https://nptel.ac.in/courses/113/105/113105024/>
3. <https://nptel.ac.in/courses/113/101/113101096/>
4. https://nptel.ac.in/content/storage2/courses/112104039/pdf_version/lecture1.pdf
