


Synthesis and structural and DNA binding studies of mono- and dinuclear copper(II) complexes constructed with —O and —N donor ligands: Potential anti-skin cancer drugs

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Mononuclear and dinuclear copper(II) complexes with thiophenecarboxylic acid, [Cu(3-TCA)₂(2,2'-bpy)] (**1**), [Cu(3-Me-2-TCA)₂(H₂O)(2,2'-bpy)] (**2**), [Cu(5-Me-2-TCA)₂(H₂O)(2,2'-bpy)] (**3**) and [Cu₂(2,5-TDCA)(DMF)₂(H₂O)₂(2,2'-bpy)₂](ClO₄)₂ (**4**) (where 3-TCA = 3-thiophenecarboxylic acid; 3-Me-2-TCA = 3-methyl-2-thiophenecarboxylic acid; 5-Me-2-TCA = 5-methyl-2-thiophenecarboxylic acid; 2,5-TDCA = thiophene-2,5-dicarboxylic acid; 2,2'-bpy = 2,2'-bipyridyl; DMF = *N,N*-dimethylformamide), were synthesized. Compounds **1–4** were extensively characterized using both analytical and spectroscopic methods. Additionally, the solid-state structures of **1** and **4** were unambiguously established from single-crystal X-ray diffraction studies. The hexacoordinated Cu(II) centre in **1** (CuO₄N₂) is a distorted octahedral geometry whereas the pentacoordinated **4** (CuO₃N₂) has distorted square pyramidal geometry. Compounds **1** and **4** exhibit intermolecular hydrogen bonding which leads to the formation of two- and three-dimensional supramolecular architectures, respectively. Spectrophotometric and computational investigations suggest that these compounds bind with DNA in minor groove binding such that $K_b = 4.9 \times 10^5 \text{ M}^{-1}$ and $K_{sv} = 3.4 \times 10^5 \text{ M}^{-1}$, and binding score of $-5.26 \text{ kcal mol}^{-1}$. The binding affinity of these complexes to calf thymus DNA is in the order **2** > **3** > **4** > **1**. Methyl-substituted thiophene ring increases the DNA binding affinity whereas unsubstituted thiophene ring DNA binding rate is reduced. The methyl group on the thiophene ring would sterically hinder π – π stacking of the ring with DNA base pairs, and subsequently they are involved in hydrophobic interaction with the DNA surface rather than partial intercalative interaction. Compounds **1–4** show pronounced activity against B16 mouse melanoma skin cancer cell lines as measured by MTT assay yielding IC₅₀ values in the micromolar concentration range. The compounds could prove to be efficient anti-cancer agents, since at a concentration as low as $2.1 \mu\text{g ml}^{-1}$ they exerted a significant cytotoxic effect in cancer cells whereas cell viability was not affected in normal cells.